

## 5,5-Dimethyl-2,3-diphenyl-5,6-dihydroimidazo[1,2-c]quinazoline

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## Key indicators

Single-crystal X-ray study

T = 296 K

Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$ 

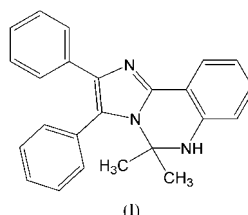
R factor = 0.039

wR factor = 0.108

Data-to-parameter ratio = 14.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound,  $\text{C}_{24}\text{H}_{21}\text{N}_3$ , was synthesized by the reaction of 4,5-diphenyl-2-(2-nitrophenyl)imidazole with acetone, induced by a low-valent titanium reagent ( $\text{TiCl}_4/\text{Zn}$ ). The dihydropyrimidine ring shows a skew-boat conformation.

## Comment

Quinazolines are an important class of compounds found in many naturally occurring products (*e.g.* hinckdentine A; Blackman *et al.*, 1987; Billimoria & Cava, 1994), and employed as potent agents (Helissey *et al.*, 1994; Brana *et al.*, 1994; Riou *et al.*, 1991; Ibrahim *et al.*, 1988). Low-valent titanium reagents have an exceedingly high ability to promote the reductive coupling of carbonyl compounds and are attracting increasing interest in organic synthesis (McMurry, 1983; Shi *et al.*, 1993, 1997, 1998, 2003). We report here the crystal structure of the title compound, (I), synthesized by the reaction of 4,5-diphenyl-2-(2-nitrophenyl)imidazole with acetone, induced by a low-valent titanium reagent ( $\text{TiCl}_4/\text{Zn}$ ).

Atoms N1, C6, C1, C8, N2 and C7 form a dihydropyrimidine ring, with an interatomic distance of 1.454 (2) Å for N1—C7 and 1.492 (2) Å for N2—C7, showing that these C—N bonds are single. The dihydropyrimidine ring adopts a skew-boat conformation; atoms C6, C1, C8 and N1 are coplanar, while atoms N2 and C7 deviate from this plane by 0.274 (1) and 0.690 (1) Å, respectively. The dihedral angle between phenyl rings C11—C16 and C17—C22 is 76.40 (2)°. In addition, because of the existence of a conjugated system, the distances N1—C6 [1.385 (2) Å], N2—C10 [1.389 (2) Å] and N2—C8 [1.371 (2) Å] are significantly shorter than the typical  $\text{Csp}^2-\text{N}$  bond distance (1.426 Å; Lorente *et al.*, 1995). The molecules are linked by an N1—H1N $\cdots$ N3<sup>i</sup> hydrogen bond (Table 2), forming zigzag chains along the *c* axis (Fig. 2).

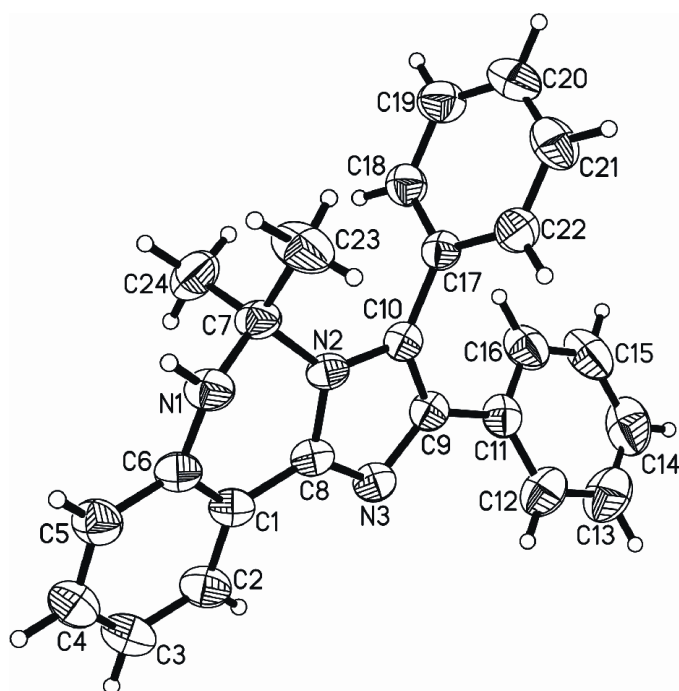
## Experimental

The title compound, (I), was prepared by the reaction of 4,5-diphenyl-2-(2-nitrophenyl)imidazole with acetone induced by a low-valent titanium reagent ( $\text{TiCl}_4/\text{Zn}$ ) (m.p. 513–514 K). Single crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution.

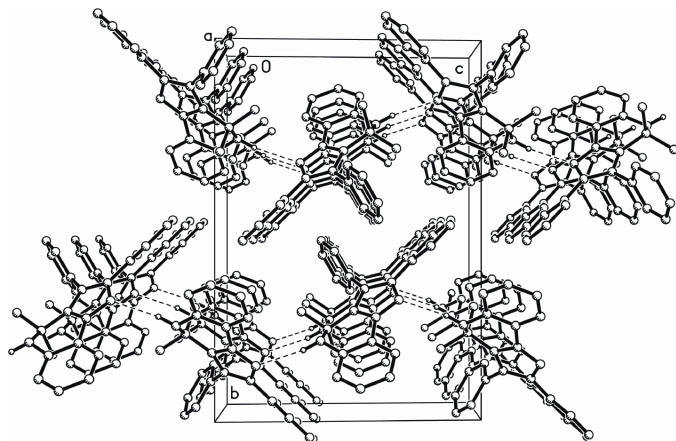
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**Figure 1**  
The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.



**Figure 2**  
The molecular packing diagram for (I).

#### Crystal data

$C_{24}H_{21}N_3$   
 $M_r = 351.44$   
 Monoclinic,  $P2_1/c$   
 $a = 8.140$  (1) Å  
 $b = 18.522$  (2) Å  
 $c = 12.963$  (2) Å  
 $\beta = 94.32$  (1)°  
 $V = 1948.8$  (3) Å<sup>3</sup>  
 $Z = 4$

#### Data collection

Siemens P4 diffractometer  
 $\omega$  scans  
 Absorption correction: none  
 4071 measured reflections  
 3527 independent reflections  
 2148 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.010$

$D_x = 1.198$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 35 reflections  
 $\theta = 3.8$ – $14.8^\circ$   
 $\mu = 0.07$  mm<sup>-1</sup>  
 $T = 296$  (2) K  
 Block, colorless  
 $0.58 \times 0.56 \times 0.50$  mm

$\theta_{max} = 25.3^\circ$   
 $h = 0 \rightarrow 9$   
 $k = 0 \rightarrow 22$   
 $l = -15 \rightarrow 15$   
 3 standard reflections  
 every 97 reflections  
 intensity decay: 3.2%

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.039$   
 $wR(F^2) = 0.108$   
 $S = 0.96$   
 3527 reflections  
 251 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0558P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} < 0.001$   
 $\Delta\rho_{max} = 0.16$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.16$  e Å<sup>-3</sup>  
 Extinction correction: *SHELXTL*  
 Extinction coefficient: 0.0177 (15)

**Table 1**

Selected geometric parameters (Å, °).

N1—C6	1.385 (2)	N3—C9	1.382 (2)
N1—C7	1.454 (2)	C1—C8	1.454 (2)
N2—C8	1.3714 (19)	C9—C10	1.369 (2)
N2—C10	1.389 (2)	C9—C11	1.480 (2)
N2—C7	1.492 (2)	C10—C17	1.482 (2)
N3—C8	1.326 (2)		
C6—N1—C7	120.62 (14)	N2—C7—C23	112.12 (16)
C8—N2—C10	107.00 (13)	N1—C7—C24	112.39 (16)
C8—N2—C7	121.77 (14)	N2—C7—C24	107.24 (14)
C10—N2—C7	130.31 (13)	N3—C8—N2	111.56 (15)
C8—N3—C9	105.11 (13)	N3—C8—C1	128.48 (15)
N1—C6—C5	122.29 (16)	N2—C8—C1	119.94 (14)
N1—C6—C1	118.19 (17)	N3—C9—C11	120.30 (14)
N1—C7—N2	105.71 (13)	C9—C10—N2	105.26 (14)
N1—C7—C23	106.98 (15)	N2—C10—C17	124.83 (13)
C7—N1—C6—C5	-150.22 (17)	C10—N2—C7—C23	-43.1 (2)
C7—N1—C6—C1	34.7 (2)	C8—N2—C7—C24	-86.93 (18)
C4—C5—C6—N1	-176.38 (17)	C9—N3—C8—N2	-1.72 (18)
C2—C1—C6—N1	176.95 (16)	C9—N3—C8—C1	176.59 (16)
C8—C1—C6—N1	-1.3 (2)	C10—N2—C8—N3	2.48 (18)
C6—N1—C7—N2	-48.1 (2)	C7—N2—C8—N3	172.58 (14)
C6—N1—C7—C23	-167.78 (17)	C6—C1—C8—N3	169.57 (16)
C6—N1—C7—C24	68.5 (2)	C6—C1—C8—N2	-12.2 (2)
C8—N2—C7—N1	33.2 (2)	C7—N2—C10—C9	-171.07 (16)
C10—N2—C7—N1	-159.28 (15)	C7—N2—C10—C17	8.3 (3)
C8—N2—C7—C23	149.38 (16)		

**Table 2**

Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1N $\cdots$ N3 <sup>i</sup>	0.883 (18)	2.260 (19)	3.133 (2)	169.5 (17)

Symmetry code: (i)  $x, \frac{3}{2} - y, z - \frac{1}{2}$ .

The H atom on nitrogen was refined isotropically, with the N—H bond length restrained to 0.88 (2) Å; other H atoms were positioned geometrically and refined as riding [C—H = 0.93–0.98 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$ ].

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Sheldrick, 1997); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*.

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